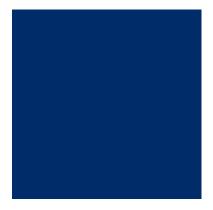
SmartFluxx SA1508SS

Nitrogen membrane module



Parker hollow-fibre membrane modules produce nitrogen gas from compressed air to offer a costeffective, reliable and safe alternative to traditional cylinder or liquid nitrogen gas supplies.

Nitrogen is used as a clean, dry, inert gas primarily for removing oxygen from products and/or processes.

Parker modules can be built into a custom-made nitrogen generator or can be integrated with your (production) process to provide an on-demand, continuous source of nitrogen gas. Gas which can be used in a wide range of industries including food, beverage, pharmaceutical, laboratory, chemical, heat treatment, electronics, transportation, oil & gas, mining and marine.



Manufacture information:

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Benefits:

- Less membrane modules needed per nitrogen system
 More nitrogen per fibre is produced from Parker hollow-fibre membranes than any other in the world
- Use of low pressure standard industrial compressor
 No high pressure compressor needed to obtain required nitrogen flow
- Energy savings
 Operation at a low pressure requires less energy
- Reduced CO₂ emissions
 No heater required to open polymer membrane structure, thus reducing the energy consumption
- Robust fibre
 Most tolerant fibre to particle contamination
- Large membrane diameter Lowest membrane module pressure drop

- Strong engineering plastic Life-expectancy of more than 10 years
- Factory membrane ageing, pre-delivery
 No performance decrease over time due to fibre ageing
- Quick start-up time
 Required nitrogen purity is produced instantly, no time needed to heat-up
- Flexible mounting arrangements
 Can be mounted horizontal
 or vertical
- Low noise operation
 Radiated noise generated by membrane technology is extremely low
- No maintenance required No user serviceable parts
- Small system footprint
 Less modules needed to
 produce nitrogen requirements



Performance data

Describe 0/	Nominal Nitrogen¹ flow rate in m³/hr² (SCFM)²				2	
Purity %	99.5	99.0	98.0	97.0	96.0	95.0
4 bar g	2.8	4.0	5.7	7.1	9.5	10.9
(58 psi g)	(1.6)	(2.4)	(3.4)	(4.2)	(5.6)	(6.4)
5 bar g	3.7	5.3	7.9	10.2	12.8	15.2
(72.5 psi g)	(2.2)	(3.1)	(4.6)	(6)	(7.5)	(8.9)
6 bar g	4.7	7.0	10.2	13.0	15.7	20.5
(87 psi g)	(2.8)	(4.1)	(6)	(7.7)	(9.2)	(12.1)
7 bar g	6.1	8.5	12.3	16.5	19.5	24.3
(101.5 psi g)	(3.6)	(5)	(7.2)	(9.7)	(11.5)	(14.3)
8 bar g	6.9	9.7	14.3	20.2	23.3	28.1
(116 psi g)	(4.1)	(5.7)	(8.4)	(11.9)	(13.7)	(16.5)
9 bar g	7.8	11.1	17.0	22.2	27.0	32.2
(130.5 psi g)	(4.6)	(6.5)	(10)	(13.1)	(15.9)	(19)
10 bar g	8.6	12.6	18.5	24.2	30.2	37.4
(145 psi g)	(5.1)	(7.4)	(10.9)	(14.2)	(17.8)	(22)
11 bar g	9.6	14.2	20.7	27.3	33.0	41.0
(159.5 psi g)	(5.7)	(8.4)	(12.2)	(16.1)	(19.4)	(24.1)
12 bar g	10.5	15.2	22.9	29.5	36.6	45.6
(174 psi g)	(6.2)	(8.9)	(13.5)	(17.4)	(21.5)	(26.8)
13 bar g	11.3	16.3	24.9	32.0	39.5	48.8
(188.5 psi g)	(6.7)	(9.6)	(14.7)	(18.8)	(23.2)	(28.7)

Describe 0/	Nominal Feed-air consumption at nitrogen flow rate in m³/hr² (SCFM)²					
Purity %	99.5	99.0	98.0	97.0	96.0	95.0
4 bar g	21	21	22	22	26	27
(58 psi g)	(12)	(12)	(13)	(13)	(15)	(16)
5 bar g	24	26	29	31	34	36
(72.5 psi g)	(14)	(15)	(17)	(18)	(20)	(21)
6 bar g	29	33	36	38	41	48
(87 psi g)	(17)	(19)	(21)	(22)	(24)	(28)
7 bar g	36	38	41	48	50	56
(101.5 psi g)	(21((22)	(24)	(28)	(29)	(33)
8 bar g	38	42	47	56	58	63
(116 psi g)	(22)	(25)	(28	(33)	(34)	(37)
9 bar g	44	48	55	62	67	72
(130.5 psi g)	(26)	(28)	(32)	(36)	(39)	(42)
10 bar g	50	56	61	68	75	84
(145 psi g)	(29)	(33)	(36)	(40)	(75)	(44)
11 bar g	51	60	66	74	80	91
(159.5 psi g)	(30)	(35)	(39)	(44)	(47)	(54)
12 bar g	57	65	76	83	92	103
(174 psi g)	(34)	(38)	(45)	(49)	(54)	(61)
13 bar g	66	72	85	92	101	113
(188.5 psi g)	(39)	(42)	(50)	(54)	(59)	(67)

Maximum pressure drop at Purity <0.2 bar Values between brackets are indicative of imperial values

For purities >99.5% please contact Parker

Ambient Conditions

Ambient temperature	+2°C to +50°C (+36°F to 122°F)
Ambient pressure	atmospheric
Air quality	clean air without contaminants

Mechanical Design Housing

Design pressure	15 bar g ⁴ (217 psi g) ⁴
Design temperature	65°C ⁴ (149°F) ⁴

⁴Membrane operating limits are lower

Operating Condtions Feed-air

Maximum operating pressure	13.0 bar g (190 psi g)
Min. / Max. operating temperature	+2°C to +50°C (+36°F to 122°F)
Maximum oil vapour content	<0.01 mg/m³ (<0.01 ppm (w))
Particles	filtered at 0.01 µm cut off
Relative humidity	<100% (non condensing)

Material

Housing	Stainless Steel
Coating	None

Flow Rate Corrections

Nitrogen flow rate at feed-air temperatures other than 20°C	Use bulletin S3.1.240 ³
Feed-air consumption at feed-air temperatures other than 20°C	Use bulletin S3.1.240 ³

³ Revision number may vary, make sure to use the most recent revision

Services Available on Request

Material certificates EN10204-3.1 on housing material (for Stainless Steel only) 3D model CAD STEP file

Weight, Dimensions and Connections

Dimensions H x Ø D	1654 x 114 mm (65.12" x 4.49")
Weight	18 kg (40 lb)
Connection feed-air	G ³ / ₄ " female to ISO 228
Connection nitrogen enriched air	G ³ / ₄ " female to ISO 228
Connection oxygen enriched air at atmospheric pressure	G 1" female to ISO 228
Dimensional drawing	Refer to K3.1.330

Note

Parker membrane systems produce both nitrogen and oxygen enriched air. Nitrogen enriched air can cause suffocation and oxygen enriched air causes increased fire hazards. The oxygen enriched air is available at ambient pressure and pressure build-up of enriched oxygen at the outlet must be prevented, otherwise a serious (reversible) decrease in performance will result. The nitrogen enriched air produced should be treated as pressurised air.

For more information please contact your local sales office or visit www.parker.com

Parker has a continuous policy of product development and although the company reserves the right to change specifications, it attemps to keep customers informed of any alterations.

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Catalogue: S3.1.248_ENe 03/14



¹ Parker membranes separate oxygen from pressurised air. The composition of the product is determined by measuring the residual oxygen content. The nitrogen content is calculated by subtracting the residual oxygen content from 100 %. Air is composed of nitrogen (78.1%), oxygen (20.9 %), Argon (0.9 %), CO₂ (0.03 %), and some trace inert gases. Remember that the value that is normally called the nitrogen content actually is the inert gas content.

 $^{^2}$ m³/hr refers to conditions at 1013 mbar(a) and 20 $^{\circ}\text{C}$